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Stereoselective formal synthesis of aspergillide A

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ABSTRACT

The stereoselective formal synthesis of aspergillide A (1), a cytotoxic 14-membered macrolide, is disclosed. The key intermediate, a trisubstituted tetrahydropyran core is prepared by Sml₂-induced intramolecular reductive cyclization as well as by using sequential α -aminooxylation, Horner–Wadsworth– Emmons olefination, and followed by Oxa-Michael cyclization. Other notable transformations in the synthesis include the use of Jacobsen's hydrolytic kinetic resolution, esterification, ring-closing metathesis (RCM), and cross-metathesis (CM) reactions.

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Recently, three novel 14-membered macrolides, named aspergillides A, B, and C (Fig. 1, 1–3) were isolated by Kusumi and coworkers¹ from marine-derived fungus *Aspergillus ostianus* strain 01F313, cultured in a 1/2PD medium containing bromine-modified artificial sea-water. These compounds were shown to exhibit significant cytotoxic activity against mouse lymphocytic leukemia cells with an IC₅₀ value of 2.1, 70, and 2.0 µg/mL, respectively. The structures originally proposed for aspergillides A (1) and B (2) were revised² by X-ray crystallography studies and the structure for aspergillide C (3) was confirmed to be correct by performing its synthesis.³ Both their unique pharmaceutical profile and challenging chemical architectures have attracted considerable interest, leading to numerous attempts and several successful syntheses of 2 and 3,⁴ but the synthesis of 1 is rare⁵ so it has driven us to take-up the synthesis of aspergillide A (1).

From a retrosynthetic perspective (Scheme 1), we planned that the side chain installation onto trisubstituted tetrahydropyran core **4** would be the late stage reaction. Thus, the 14-membered macrolide **1** can be achieved from two fragments, **4** and **5**, via esterification followed by a ring-closing metathesis reaction or cross-metathesis reaction and Yamaguchi lactonization. The trisubstituted tetrahydropyran core **4** was prepared by two different routes using a Sml₂-induced intramolecular reductive cyclization or sequential α -aminooxylation followed by in situ Horner–Wadsworth–Emmons olefination and Oxa-Michael reaction from **6** and **7**, respectively. The **6** in turn could be prepared from the known epoxide **8**, whereas the aldehyde **7** could be obtained from the epoxide **9**.

In route A (Scheme 2), our synthesis commenced from chiral epoxide **8**, prepared from 4-pentene-1-ol following the literature procedure.⁶ Opening of epoxide⁷ **8** using trimethylsulfonium iodide, *n*-BuLi in dry THF at $-10 \,^{\circ}$ C provided secondary allylic alcohol **10** in 75% yield. The plan for the diastereoselective construction of **13** was based on Sml₂ reductive cyclization^{8,9} of aldehydes such as **6**. To this end, the allylic alcohol **10** was condensed with methyl propiolate. This was achieved by slow addition of methyl propiolate¹⁰ to compound **10** via syringe pump over 16 h to deliver the intermediate **11** in 90% isolated yield. Next, deprotection of the THP ether **11** by using PTSA in MeOH released the primary alcohol **12**, which was immediately oxidized using IBX in DMSO/CH₂Cl₂ to provide aldehyde **6**. As anticipated, the Sml₂-mediated cyclization of **6** proceeded to give **13** with high diasterocon-



Aspergillide A (1) Aspergillide B (2) revised structures

Figure 1.

Aspergillide C (3)

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Scheme 2. Reagents and conditions: (a) trimethylsulfonium iodide, *n*-BuLi, THF, -10 °C, 4 h, 75%; (b) methyl propiolate, DMAP, CH₃CN, rt, 16 h, 90%; (c) PTSA, MeOH, 0 °C to rt, 12 h, 81%; (d) IBX (*o*-iodoxybenzoic acid), DMSO, CH₂Cl₂, 0 °C to rt, 1 h, 78%; (e) Sml₂ (0.1 M solution) (3 equiv), MeOH (3 equiv), THF, 0 °C, 1 h, 92%; (f) TBSCl, imidazole, CH₂Cl₂, 0 °C to rt, 20 h, 88%; (g) LiOH, MeOH/H₂O (8:2), 0 °C to rt, 10 h, 78%.

trol. Thus treatment of **6** with 3 equiv of Sml_2^{11} in the presence of 3 equiv of dry MeOH in THF effected reductive cyclization to give 2,6-*syn*-2,3-*anti*-tetrahydropyran **13** in 92% yield as the single product. The hydroxy group was protected as TBS ether **14** and cleavage of the methyl ester furnished acid **4** in excellent yield.

Product **14** was thoroughly characterized with the help of 2-D nuclear Overhauser effect spectroscopy (NOESY) and double quantum filter correlation spectroscopy (DQF-COSY) experiments. J_{H2-H3} = 9.1 Hz, J_{H5-H6} = 12.2, and $J_{H5'-H6}$ = 1.7 Hz suggest that the H-2 and H-6 are diaxially disposed with respect to each other, which is further confirmed by exclusive, strong NOE cross-peaks between H2 and H6. Hence *R*- and *S*-stereocenter for C2 and C3, respectively. Further NOE correlation H2/H4 and H4/H6 is consistent with the chair conformation for six-membered pyran ring. The minimum energy structure is adequately supported by NMR data. These data (Figs. 2 and 3) suggest that the structure of the newly formed ring in **14**¹² is the desired 2,6-*syn*-2,3-*anti* tetrahydropyran.

In route B (Scheme 3), the synthesis of the key intermediate 13 started from the epoxy alcohol 9 (98% ee, measured by chiral HPLC) prepared as reported in our Letter.¹³ The alcohol was treated with TPP and I_2 in Et₂O/CH₃CN (3:1) in the presence of imidazole to get the 2,3-epoxy iodide 15 in good yield, which on refluxing in MeOH with Zn yielded the allylic alcohol 16. The allylic alcohol was protected as TBS ether 17 followed by removal of PMB group that provided the primary alcohol 18, which on oxidation using IBX in DMSO/CH₂Cl₂ afforded the aldehvde. The aldehvde without isolation was subjected to the crucial MacMillan¹⁴ α -aminooxylation using nitrosobenzene and 20 mol % p-proline in DMSO followed by in situ Horner-Wadsworth-Emmons olefination with trimethyl phosphonoacetate and Cs_2CO_3 as base to furnish γ -anilinoxy- α , β unsaturated ester **19a**, which was further treated with Cu(OAc)₂ in ethanol at room temperature to cleave the O-N bond providing the ester **19** with high enantiopurity (dr, >95:<5). Exposure of compound 19 to TBAF for 20 h effected a silyl group removal and a

smooth Oxa-Michael cyclization under thermodynamic conditions in one-pot provided exclusively the 2,6-*syn*-2,3-*anti* tetrahydropyran **13** (70%). The hydroxy group was protected as TBS ether **14** and cleavage of the methyl ester furnished acid **4** in excellent yield. The analytical and optical rotation values of **13**, **14**, and **4** exactly matched with those of **13**, **14**, and **4** synthesized by Scheme 2.

Now, the stage is set to the synthesis of **1** to fasten the desired fragments **4** and the known alcohol **5**¹⁵ together to obtain the 14membered macrocycle **1** (Scheme 4). Thus, the esterification of alcohol **5** with carboxylic acid **4** was carried out under Yamaguchi conditions¹⁶ to produce compound **20** in 79% yield. This set the stage for the macrocyclization by ring-closing metathesis.

Reaction of diene **20** with second or first generation Grubbs' catalyst¹⁷ in refluxing dichloromethane or using Grubbs–Hoveyda (G–H) catalyst in the presence of 1,4-benzoquinone¹⁸ also resulted in the exclusive formation of *Z*-isomer **21** in 90% yield. Finally, the (*Z*)-isomer **21** was subjected to desilylation under TBAF conditions to give the macrocyclic core **22** (J = 10.5 Hz) in 85% yield. The structure of macrolactone **22** was fully characterized by ¹H NMR, ¹³C NMR, and mass spectral data.

In addition, the cross-metathesis of hydroxy acid **4** with alcohol **5** carried out in the presence of 10 mol% of Grubbs-II catalyst afforded **23** in 15 min as a mixture of E/Z in a ratio of 9:1. The formation of the product was confirmed by ¹H NMR analysis and comparison with the literature data.^{5a}

It is noteworthy to mention that when our work was under progress, a note appeared^{5b} on the synthesis of aspergillide A **1** indicating similar results on RCM and cross-metathesis reactions. The spectral and analytical (¹H, ¹³C NMR, mass, IR, and optical rotation) data of our synthetic materials **22** and **23** were in complete agreement with those reported.^{5a,5b} As per the literature, the macrolactonization of **23** proved to be a difficult task. Only in the case of TBS-ether and MOM-ether the product was obtained in 30% and 20% yields, respectively, under Yamaguchi conditions.^{5a,5c} The seco



Figure 2. Expansion of the NOESY spectrum showing the characteristic NOE correlations.



Figure 3. Chemical structure and energy-minimized structure of 14.



Scheme 3. Reagents and conditions: (a) TPP, imidazole, iodine, Et₂O/CH₃CN (3:1), 0 °C, 1 h; (b) Zn, MeOH, reflux, 1 h, (92% overall from **9**); (c) TBSCl, imidazole, CH₂Cl₂, 0 °C to rt, 4 h, 96%; (d) DDQ, CH₂Cl₂/H₂O (9:1), 0 °C, 1 h, 91%; (e) IBX (o-iodoxybenzoic acid), DMSO, CH₂Cl₂, 0 °C to rt, 10 h; (f) (i) PhNO, p-proline, DMSO, rt, 15 min, (ii) trimethyl phosphonoacetate, cesium carbonate, rt, 2 h (60% overall from **18**); (g) Cu(OAc)₂, EtOH, rt, overnight, 85%; (h) TBAF, THF, 20 h, rt, 70%; (i) TBSCl, Imidazole, CH₂Cl₂, 0 °C to rt, 20 h, 88%; (j) LiOH, MeOH/H₂O (8:2), 0 °C to rt, 10 h, 78%.

acid bearing benzyl ether as per the report is not stable under any macrolactonization conditions.^{5b} Since the conversion of **22** and **23** to aspergillide A **1** has already been reported in the literature,^{5a,5b} the present sequence herein constitutes a formal synthesis of aspergillide A **1**.

In conclusion, a formal synthesis of the 14-membered macrolide, aspergillide A (1) has been demonstrated. This synthesis features a key SmI₂ reductive cyclization step and sequential α -aminooxylation–Horner–Wadsworth–Emmons olefination and oxa-Michael cyclization reactions to access the trisubstituted pyr-



Scheme 4. Reagents and conditions: (a) DCC, DMAP, CH_2Cl_2 , 0 °C to rt, 6 h, 79%; (b) Ru-I, CH_2Cl_2 , reflux, 2 h, 90%; (c) TBAF, THF, rt, overnight, 85%; (d) Ru-II, CH_2Cl_2 , reflux, 15 min, 42%.

an core. Other salient features of the approach include the use of ring-closing metathesis and cross-metathesis reactions. We believe that this approach sets the stage not only for the total synthesis of aspergillide A (1) but also entry to a diversity of analogues through the installation of various side chains. Studies in this direction are underway.

Acknowledgment

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2010.06.013.

References and notes

- 1. Kito, K.; Ookura, R.; Yoshida, S.; Namikoshi, M.; Ooi, T.; Kusumi, T. Org. Lett. 2008, 10, 225–228.
- 2. Ookura, R.; Kito, K.; Saito, Y.; Kusumi, T.; Ooi, T. Chem. Lett. 2009, 38, 384.
- 3. Nagasawa, T.; Kuwahara, S. Org. Lett. 2009, 11, 761-764.
- Syntheses of aspergillide B and C: (a) Hande, S. M.; Uenishi, J. Tetrahedron Lett. 2009, 50, 189–192; (b) Liu, J.; Xu, K.; He, J.-M.; Zhang, L.; Pan, X.-F.; She, X.-G. J. Org. Chem. 2009, 74, 5063–5066; (c) Nagasawa, T.; Kuwahara, S. Biosci. Biotechnol. Biochem. 2009, 73, 1893–1894; (d) Diaz-Oltra, S.; Angulo-Pachon, C. A.; Kneeteman, M. N.; Murga, J.; Carda, M.; Marco, J. A. Tetrahedron Lett. 2009, 50, 3783–3785; (e) Panarese, J. D.; Waters, S. P. Org. Lett. 2009, 11, 5086–5088.
- (a) Nagasawa, T.; Kuwahara, S. *Tetrahedron Lett.* **2010**, *51*, 875–877; (b) Díaz-Oltra, S.; Angulo-Pachon, C. A.; Murga, J.; Carda, M.; Marco, J. A. J. Org. Chem. **2010**, 75, 1775–1778; (c) Fuma, H.; Yamaguchi, H.; Sasaki, M. Org. Lett. **2010**, *12*, 1848–1851.
- (a) Schwartz, N. N.; Blumbergs, J. H. J. Org. Chem. **1964**, 29, 1976; (b) Schaus, S. E.; Brandes, B. D.; Larrow, J. F.; Tokunaga, M.; Hansen, K. B.; Gould, A. E.; Kurrow, M. E.; Jacobsen, E. N. J. Am. Chem. Soc. **2002**, *124*, 1307.
- 7. Alcaraz, L.; Harnett, J. J.; Mioskowski, C.; Martel, J. P.; Legall, T.; Shin, D.-S.; Falck, J. R. Tetrahedron Lett. **1994**, 35, 5449.
- (a) Hori, N.; Matsukura, H.; Matsuo, G.; Nakata, T. Tetrahedron Lett. **1999**, 40, 2811; (b) Hori, N.; Matsukura, H.; Nakata, T. Org. Lett. **1999**, 1, 1099; (c) Matsuo, G.; Hori, N.; Nakata, T. Tetrahedron Lett. **1999**, 40, 8859; (d) Matsuo, G.; Kadohama, H.; Nakata, T. Chem. Lett. **2002**, 148–149; (e) Hori, N.; Matsukura, H.; Matsuo, G.; Nakata, T. Tetrahedron **2002**, 58, 1853.
- Selected papers: (a) Fuwa, H.; Ebine, M.; Sasaki, M. J. Am. Chem. Soc. 2006, 128, 9648; (b) Fuwa, H.; Kakinuma, N.; Tachibana, K.; Sasaki, M. J. Am. Chem. Soc. 2002, 124, 14983; (c) Kadota, I.; Takamura, H.; Sato, K.; Ohno, A.; Matsuda, K.; Satake, M.; Yamamoto, Y. J. Am. Chem. Soc. 2003, 125, 11893; (d) Nagumo, Y.; Oguri, H.; Shindo, Y.; Sasaki, S.; Oishi, T.; Hirama, M.; Tomioka, Y.; Mizugaki, M.; Tsumuraya, T. Bioorg. Med. Chem. Lett. 2001, 11, 2037; (e) Matsuo, G.; Kawamura, K.; Hori, N.; Matsukura, H.; Nakata, T. J. Am. Chem. Soc. 2004, 126, 14374; (f) Takahashi, S.; Kubota, A.; Nakata, T. Angew. Chem., Int. Ed. 2002, 41, 4751.
- (a) Winterfeldt, E. Chem. Ber. 1964, 97, 1952–1958; (b) Winterfeldt, E.; Preuss, H. Chem. Ber. 1966, 99, 450–458.
- (a) Namy, J. L.; Girard, P.; Kagan, H. B. Nouv. J. Chim. **1977**, 1, 5; (b) Girard, P.; Namy, J. L.; Kagan, H. B. J. Am. Chem. Soc. **1980**, 102, 2693; (c) Kagan, B. H. New J. Chem. **1990**, 14, 453.

- 12. Spectral data: methyl 2-((2R,3S,6R)-3-[1-(tert-butyl)-1,1-dimethylsilyl]oxy-6-vinyltetrahydro-2H-2 pyranyl)acetate (14): $[\alpha]_D^{25} + 24.3$ (c 0.01 g/mL, CHCl₃); ¹H NMR: (CDCl₃, 600 MHz): δ 5.81 (ddd, J = 5.1, 10.7, 17.4 Hz, 1H), 5.20 (td, J = 1.6, 17.4 Hz, 1H), 5.06 (td, J = 1.6, 17.4 Hz, 1H), 3.68 (s, 3H), 3.65 (dt, J = 3.2, 9.1 Hz, 1H), 3.35 (ddd, J = 4.4, 9.1, 10.3 Hz, 1H), 2.82 (dd, J = 3.3, 15.2 Hz, 1H), 2.39 (dd, J = 9.2, 15.2 Hz, 1H), 2.02 (qd, J = 3.8, 12.6 Hz, 1H), 1.77 (qd, J = 2.8, 12.6 Hz, 1H), 1.62–1.41 (m, 2H), 0.88 (s, 9H), 0.06 (s, 3H), 0.05 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz): 172.2, 138.3, 114.8, 790, 77.6, 70.6, 51.5, 37.9, 33.2, 30.9, 25.7, 17.8, -4.0, -4.8; IR (Neat): 3060, 2933, 2858, 1744, 1640, 1464, 1437, 1342, 1194, 1095, 990,927, 897, 840, 775, 671 cm⁻¹; LC-MS: 315 [M+H]*; HMS: m/z [M+Na]* calcd for C1₁₆H₃₁O₄Si: 315.1991; found: 315.1991.
- Sabitha, G.; Bhaskar, V.; Yadagiri, K.; Yadav, J. S. Synth. Commun. 2007, 37, 2491–2500.
- (a) Mangion, I. K.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 3696–3697;
 (b) Brown, S. P.; Brochu, M. P.; Sinz, C. J.; MacMillan, C. W. C. J. Am. Chem. Soc. 2003, 125, 10808–10809;
 (c) Zhong, G. Angew. Chem., Int. Ed. 2003, 42, 4247;
 (d) Hayashi, Y.; Yamaguchi, J.; Hibino, K.; Shoji, M. Tetrahedron Lett. 2003, 44, 8293;
 (e) Zhong, G. Chem. Commun. 2004, 606–607.
- Dixon, D. J.; Ley, S. V.; Tate, E. W. J. Chem. Soc., Perkin Trans. 1 2000, 2385–2394.
 Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. Bull. Chem. Soc. Jpn.
- **1979**, 52, 1989–1993. 17. (a) Fürstner, A.; Nevado, C.; Waser, M.; Tremblay, M.; Chevrier, C.; Teplý, F.; Airca C.; Moulin, F.; Müllar, O. J. Am. Cham. Soc. **2007**, 120, 0150, 0151.
- Aïssa, C.; Moulin, E.; Müller, O. J. Am. Chem. Soc. 2007, 129, 9150–9161; (b) Garbaccio, R. M.; Danishefsky, S. J. Org. Lett. 2000, 20, 3127–3129.
 Hong S. H.; Sandari D. B.; Leo G. W.; Grinbia P. H. L. Am. Chem. Soc. 2005, 127.
- Hong, S. H.; Sander's D. P.; Lee, C. W.; Grubbs, R. H. J. Am. Chem. Soc. 2005, 127, 17160.

Selected spectral data: (3*R*)-6-(tetrahydro-2*H*-2-pyranyloxy)-1-hexen-3-ol (**10**): $[\alpha]_{25}^{25}$ -6.2 (c 0.01 g/mL, CHCl₃): ¹H NMR (CDCl₃, 300 MH2): δ 5.91–5.78 (m, 1H), 5.22 (d, *J* = 17.1 Hz, 1H), 5.07 (d, *J* = 10.3 Hz, 1H), 4.60–4.56 (m, 1H), 4.16–4.08 (m, 1H), 3.87–3.72 (m, 2H), 3.53–3.44 (m, 1H), 3.43–3.34 (m, 1H), 2.23 (br s, 1H, OH), 1.89–1.47 (m, 10H); ¹³C NMR (CDCl₃, 75 MHz): 141.0, 114.4, 98.8, 72.7, 67.5, 62.2, 34.2, 30.5, 25.8, 25.3, 19.4; IR (Neat): 3430, 3060, 2941, 2866, 1640, 1442, 1352, 1266, 1200, 1123, 1071, 1027, 988, 913, 868, 810, 758, 669 cm⁻¹; ESIMS: *m*/2 223 [M+Na]⁺; HRMS: *m*/z [M+Na]⁺ calcd for C₁₁H₂₀O₃Na: 223.1310; found: 223.1303.

 $\begin{array}{l} \mbox{Methyl (E)-3-[(1R)-1-(3-hydroxypropyl)-2-propenyl]oxy-2-propenote (12): <math display="inline">[\alpha]_D^{25} \\ -5.3 \ (c \ 0.009 \ g/mL, \ CHCl_3); \ ^1H \ NMR \ (CDCl_3, \ 300 \ MHz): \ \delta \ 7.46 \ (d, \ J=12.0 \ Hz, \ 1H), \ 5.82-5.69 \ (m, \ 1H), \ 5.32-5.19 \ (m, \ 3H), \ 4.39-4.31 \ (m, \ 1H), \ 3.70-3.61 \ (m, \ 5H), \ 1.85-1.54 \ (m, \ 4H); \ ^{13}C \ NMR \ (CDCl_3, \ 75 \ MHz): \ 168.4, \ 161.6, \ 136.2, \ 118.2, \ 97.6, \ 83.7, \ 62.3, \ 51.1, \ 31.2, \ 28.1; \ IR \ (Neat): \ 3431, \ 3055, \ 3020, \ 2947, \ 1705, \ 1635, \ 1438, \ 1334, \ 1292, \ 1199, \ 1139, \ 1054, \ 967, \ 900, \ 832, \ 754 \ cm^{-1}; \ LC-MS: \ m/z \ 223 \ M+Na]^* \ calcd \ for \ C_{10}H_{16}O_4Na: \ 223.0946; \ found: \ 223.0940. \end{array}$

 $\begin{array}{l} ((2R,3R)-3.4-[(4-Methoxybenzyl)oxy]butyloxiran-2-yl)methanol (9): [x]_D^{25} +11.1 (c 0.0135 g/mL, CHCl_3); ^1H NMR (CDCl_3, 500 MHz): δ7.19 (d, J = 8.6 Hz, 2H), 6.82 (d, J = 8.6 Hz, 2H), 4.39 (s, 2H), 3.83 (dd, J = 1.9, 12.5, Hz, 1H), 3.79 (s, 3H), 3.57 (dd, J = 4.8, 12.5 Hz, 1H), 3.41 (t, J = 6.7 Hz, 2H), 2.91-2.67 (m, 1H), 2.86-2.82 (m, 1H), 1.66-1.46 (m, 6H); ^{13}C NMR (CDCl_3, 75 MHz): 159.0, 130.4, 129.1, 113.6, 72.4, 69.6, 61.6, 58.4, 55.8, 55.2, 31.2, 29.3, 22.6 ; IR: 3424, 2932, 2860, 1611, 1513, 1460, 1248, 1096, 1032, 822, 757 cm^{-1}; LC-MS: 289 [M+Na]*; RRMS: m/z [M+Na]* calcd for C15H22O4Na: 289.1415; found: 289.1407.$

 $\begin{array}{l} \label{eq:24} Methyl (2E,4S,7R)-7-[1-(tert-butyl)-1,1-dimethylsilyl]oxy-4-hydroxy-2,8-nonadienoate \\ (19): $[\alpha]_{2}^{25}$-28.4 (c~0.0225~g/mL, CHCl_3); $^{1}H~NMR (CDCl_3, 300~MHz); $^{2}6.94 (dd, $^{J}=4.5, 15.8~Hz, 1H), 6.07 (dd, $^{J}=2.2, 15.8~Hz, 1H), 5.86-5.72 (m, 1H), 5.21-5.05 (m, 2H), 4.35-4.19 (m, 2H), 3.75 (s, 3H), 2.80 (br s, 1H, OH), 1.75-1.55 (m, 4H), 0.91 (s, 9H), 0.08 (s, 3H), 0.06 (s, 3H); $^{13}C~NMR (CDCl_3, 75~MHz); 167.1, 150.4, 140.5, 119.7, 114.5, 73.5, 71.0, 51.5, 33.8, 31.8, 25.8, 18.2, -4.4, -4.9 ; IR (Neat): 3446, 2925, 2855, 1726, 1656, 1462, 1258, 1170, 1077, 923, 836, 775 cm^{-1}; LC-MS: 337 [M+Na]^{+}; HRMS: $m/z [M+Na]^{+} calcd for $C_{16}H_{30}O_4NaSi: 337.1811; found: 337.1799. \end{array}$

 $\begin{array}{l} (1R,SS,11R,14S)-14-Hydroxy-5-methyl-4,15-dioxabicyclo-[9.3.1]pentadec-9(Z)-en-3-one ($ **22** $): <math>[\alpha]_D^{25}$ +38.2 (c 0.006 g/mL, CHCl_3), $[\mathrm{it}_D^{5b}\ [\alpha]_D^{25}$ +42.9 (c 0.077, CHCl_3); $^{1}\mathrm{H}\ \mathrm{NMR}\ (\mathrm{CDCl}_3,\ 300\ \mathrm{MHz}):\ \delta\ 5.64-5.53\ (\mathrm{m},\ 1\mathrm{H}),\ 5.16\ (\mathrm{dd},\ J=2.2,\ 10.5\ \mathrm{Hz},\ 1\mathrm{H}),\ 5.06-4.93\ (\mathrm{m},\ 1\mathrm{H}),\ 4.08-4.00\ (\mathrm{m},\ 1\mathrm{H}),\ 3.56-3.47\ (\mathrm{m},\ 1\mathrm{H}),\ 3.35-3.25\ (\mathrm{m},\ 1\mathrm{H}),\ 5.06-4.93\ (\mathrm{m},\ 1\mathrm{H}),\ 4.08-4.00\ (\mathrm{m},\ 1\mathrm{H}),\ 3.56-3.47\ (\mathrm{m},\ 1\mathrm{H}),\ 3.35-3.25\ (\mathrm{m},\ 1\mathrm{H}),\ 2.32-2.09\ (\mathrm{m},\ 2\mathrm{H}),\ 2.30\ (\mathrm{t},\ J=1.3\ \mathrm{Hz},\ 1\mathrm{H}),\ 2.32-2.09\ (\mathrm{m},\ 2\mathrm{Hz}),\ 1.25\ (\mathrm{d},\ J=6.0\ \mathrm{Hz},\ 3\mathrm{Hz}),\ 1.25\ (\mathrm{d},\ J=6.0\ \mathrm{Hz},\ 3\mathrm{Hz})$

2-(2R,3S,6R)-3-hydroxy-6-[(E,6S)-6-hydroxy-1-heptenyl]tetrahydro-2H-2-pyranyl acetic acid (**23**): [α]_D²⁵ +48.4 (c 0.005 g/mL, CHCl₃), ¹H NMR (CDCl₃, 500 MHz): δ 5.67–5.52 (m, 1H), 5.39 (dd, *J* = 5.2, 15.6 Hz, 1H), 3.85–3.71 (m, 2H), 3.60–3.53 (m, 1H), 3.36–3.28 (m, 1H), 2.77 (dd, *J* = 3.1, 14.5 Hz, 1H), 2.39–2.33 (m, 1H), 2.15–1.95 (m, 4H), 1.75–1.60 (m, 1H), 1.57–1.34 (m, 5H), 1.16 (d, *J* = 6.2 Hz, 3H), 0.88 (s, 9H), 0.06 (s, 6H); LC–MS: 409 [M+Na]⁺; HRMS: *m/z* [M+Na]⁺ calcd for C₂₀H₃₈O₅NaSi: 409.5902; found: 409.5893.